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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.003 Å R factor = 0.038 wR factor = 0.100 Data-to-parameter ratio = 13.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. N⁴-Acetylsulfamerazine

In the title compound [systematic name: 4-acetamido-N-(4-methylpyrimidin-2-yl)benzenesulfonamide], $C_{13}H_{14}N_4O_3S$ (NSMZ), which is the acetyl derivative of 4-amino-N-(4-methylpyrimidin-2-yl)benzenesulfonamide (sulfamerazine, SMZ), a network of $N-H\cdots N$ and $N-H\cdots O$ interactions results in sheets of molecules in the crystal structure.

Comment

4-Amino-*N*-(4-methylpyrimidin-2-yl)benzenesulfonamide (sulfamerazine, SMZ) is commonly used as an antibacterial agent in veterinary practice. The title compound, N^4 -acetylsulfamerazine, NSMZ, (I), is the acetylation metabolite of SMZ (Nouws *et al.*, 1988, 1989). NSMZ is also the precursor of SMZ in the synthetic step before hydrolysis (Roblin & Winneck, 1940). Due to the similarity between the molecular structures of SMZ and NSMZ, NSMZ may substitute for some SMZ molecules in the crystal structure of SMZ and distrupt the hydrogen-bonding interactions with the incoming SMZ molecules. This may explain why a trace amount of NSMZ can stablize the metastable crystalline form of SMZ in an acetonitrile suspension (Gu *et al.*, 2002).



The fully ordered crystal structure of compound (I) is reported here. It crystallizes in space group C2/c with one molecule in the asymmetric unit (Fig. 1). In the crystal structure of (I), two NSMZ molecules related by a center of symmetry form a dimer *via* a pair of intermolecular $N-H\cdots N$ hydrogen bonds (Table 1), The dimers are further linked with each other by $N-H\cdots O$ intermolecular hydrogen bonds among the acetylamido groups, forming two dimensional infinite zigzag-shaped sheets parallel to the ($\overline{102}$) plane. Such sheets stack to form the whole complete crystal structure with no hydrogen-bonding interactions between them. One of the sheets in shown in Fig. 2.

Experimental

NSMZ was synthesized as suggested by Roblin & Winneck (1940). 1.06 g (9.8 mmol) of 2-amino-4-methylpyrimidine was suspended in 2.5 ml of dry pyridine. 2.38 g (10.2 mmol) of *N*-acetylsulfanilyl chloride was added gradually with stirring. The temperature of the

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reaction system was maintained below 328 K. The mixture was then heated on a steam bath for two h. A solution containing 0.44 g (11 mmol) of sodium hydroxide in 2.2 ml of water was added slowly. The pyridine was removed by distillation under reduced pressure. Yellow-brown NSMZ solid was separated by filtration. Crystals of (I) were prepared from a tetrahydrofuran (THF) solution by slow evaporation at room temperature.

Z = 8

 $D_x = 1.384 \text{ Mg m}^{-3}$

0.25 \times 0.15 \times 0.10 mm

10158 measured reflections

2596 independent reflections 2046 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation $\mu = 0.24 \text{ mm}^{-1}$

T = 173 (2) K

Block, vellow

 $R_{\rm int} = 0.035$ $\theta_{\rm max} = 25.0^{\circ}$

Crystal data

 $\begin{array}{l} C_{13}H_{14}N_4O_3S\\ M_r = 306.34\\ \text{Monoclinic, } C2/c\\ a = 12.2826 \ (14) \text{ Å}\\ b = 9.2479 \ (11) \text{ Å}\\ c = 26.010 \ (3) \text{ Å}\\ \beta = 95.610 \ (2)^\circ\\ V = 2940.2 \ (6) \text{ Å}^3 \end{array}$

Data collection

Bruker SMART 1000 CCD diffractometer ω scans Absorption correction: multi-scan *SADABS* (Bruker, 2000) $T_{\rm min} = 0.921, T_{\rm max} = 0.980$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0472P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	+ 2.453 <i>P</i>]
$wR(F^2) = 0.100$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} < 0.001$
2596 reflections	$\Delta \rho_{\rm max} = 0.32 \text{ e} \text{ \AA}^{-3}$
198 parameters	$\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected torsion angles (°).

C3-N1-C2-C1	-175.4 (2)	S1-N2-C9-N4	4.6 (3)
N2-S1-C6-C5	-107.95 (18)	S1-N2-C9-N3	-175.67 (15)

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$N1-H1A\cdotsO1^{i}$	0.8799 (10)	2.228 (11)	3.036 (2)	153 (2)
$N2-H2B\cdots N3^{ii}$	0.8799 (11)	2.014 (5)	2.885 (3)	170 (2)
Symmetry codes: (i)	$-r = \frac{1}{2}v = \frac{1}{2} = 7$	$+\frac{1}{2}$ (ii) $-r + \frac{1}{2} - \frac{1}{2}$	$v + \frac{3}{2} - 7 + 1$	1/0 (2)

The N-bound H atoms were located in difference maps and their positions were refined with the distance restraint N-H = 0.88 (1) Å and $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm N})$. The C-bound H atoms were placed in calculated positions with C-H = 0.95–0.98 Å and and refined as riding with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ or $1.5U_{\rm eq}({\rm methyl}\ {\rm C})$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT-Plus* (Bruker, 2003); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: Mercury (Version 1.4.1; Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXTL* (Bruker, 2000).



Figure 1 View of (I), showing 50% displacement ellipsoids.



Figure 2

Part of an infinite sheet formed by hydrogen-bonding (dashed lines) interactions in the crystal structure of (I)

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